

10/521989

**MUCOADHESIVE COMPOSITION AND FORMULATION FOR  
SOLUBILIZATION OF INSOLUBLE DRUGS AND PREPARATION  
METHOD THEREOF**

5

**[TECHNICAL FIELD]**

The present invention relates to a novel mucoadhesive composition for solubilization of insoluble drugs; its formulation including pharmaceutical compounds; and the preparation methods thereof, wherein said solubilizing composition is composed of 4~90% by weight of at least one selected from 10 the monoglycerides and 0.01~90 % by weight of at least one oil. The present invention also relates to a novel mucoadhesive composition including emulsifiers for solubilization of insoluble drugs; its formulation including pharmaceutical compounds; and the preparation methods thereof wherein said solubilizing composition including emulsifiers is composed of 4~90% by 15 weight of at least one selected from the monoglycerides, 0.01~90 % by weight of at least one oil, and 0.01 ~ 90% by weight of at least one selected from the emulsifiers. The compositions of the present invention are suitable as drug delivery systems since they exist as mucoadhesive liquid at physiological temperatures even though they exist as liquid or semi-solid at 20 room temperature.

**[BACKGROUND ART]**

Solubilization process is a very important step in preparing the delivery systems of insoluble drugs. To solubilize insoluble drugs, a variety

of compositions including fats, lipids and oils have been prepared in the past. These compositions, however, are decomposed by lipase in the intestine or solubilized by bile salts to form mixed micelles resulting in lowered absorption of the encapsulated drugs. To overcome these problems and to 5 increase the absorption rate, nano-sized lipid particles have also been prepared by the aid of emulsifiers since the particles of small size were absorbed through the intestinal cells easily. The present inventors have found, however, that the oily compositions that are mucoadhesive and can dissolve insoluble drug can help increase the absorption rate of the 10 encapsulated drug when taken orally even if the oily composition cannot be dispersed in water homogeneously.

Since mucoadhesive drug delivery systems can be adsorbed on the intestinal absorptive cells via oral, buccal or intranasal administration and slowly release the encapsulated drugs in the vicinity of the site of absorption, 15 drug absorption rate can be increased when taken orally or applied directly on the wound.

Conventionally used mucoadhesive drug delivery systems are mainly polymeric materials including DEAE dextran, polycarbophil, sodium alginate, hydroxypropyl methylcellulose (HPMC) and Carbopol 934 (BF Goodrich, 20 USA). Among lipids, monoglycerides are known to have high mucoadhesiveness. The mucoadhesiveness of the monoglycerides is the highest when they exist as the precursors of cubic or hexagonal phases. Among these precursors, Elyzol gel for the treatment of periodontal disease comprising metronidazole benzoate, monoglyceride and unsaturated 25 triglycerides with a small amount of water (20 %) is commercially available [Norling et. al., Formulation of a drug delivery system based on a mixture of

monoglycerides and triglycerides for use in the treatment of periodontal disease [1992] J. Clin. Periodontol. vol. 19, page 687-692]. Mucoadhesive hexagonal liquid crystalline phase is formed inside the periodontal pocket when Elyzol is injected.

5       The precursors, however, become mucoadhesive only when they come in contact with mucosal cells. If the precursors meet intestinal fluid and become hexagonal or cubic phase before contacting mucosal cells, they lose the mucoadhesiveness to a great extent. Even if the precursor reaches the mucosal cells before absorbing intestinal fluid, they can be degraded by  
10      the intestinal enzymes. Also, the formed cubic or hexagonal phase covers only limited areas of the intestine, drug is absorbed at the site of attachment only. This problem arises since the cubic phase that monoglyceride and water forms has a very high viscosity and does not migrate to lower parts of the intestine. The composition of the present invention, however, contains  
15      oils of low viscosity that helps the composition to flow inside the intestine and to coat the interior of intestine. Therefore, the amount of drug absorption per unit contact area between mucosal cells and the composition increases when the whole intestine is considered.

20      The drug content also increases since the oils with low viscosity and high solubility for lipophilic drugs are included in the composition. For example, the solubility of pyrene, a model drug, in tricaprylin, a saturated triglyceride, is 92.9 mg/ml, whereas that in monoolein is 43.6 mg/ml. By adding tricaprylin in the composition, more pyrene can be solubilized. Since the composition of the present invention does not contain water, the  
25      composition is stable for a long period of time without undergoing oxidation and hydrolysis of the components.

Monoolein is mucoadhesive when it exists as a low viscous liquid. The melting point of monoolein of high purity (99.5 % pure) is 37 °C and that of Myverol 18-99 (Danisco, Denmark) is 35 ~ 40 °C. The melting point of monoolein is similar to the body temperature, and it can absorb the stomach fluid or intestinal fluid if the soft capsule containing monoolein dissolves in stomach or intestine, respectively. Since the cubic phase that monoolein forms upon absorbing water is a highly viscous gel at 37 °C, it only coats a limited contact area. The present invention, on the other hand, forms a low viscosity phase that can coat the intestinal cells evenly providing a wider area for drug absorption.

If the drug is encapsulated in the oil without monoglyceride, the composition is adsorbed on the intestine momentarily, but the drug absorption is ineffective since the oil has to be digested before being absorbed. Monoglyceride, however, can be absorbed directly without being digested on the mucosal cells and therefore can carry the drug with it. Therefore, the composition of the present invention containing a mixture of monoglyceride and oil can coat a wide surface area of intestine, can load drug at a high concentration and can help absorption of the drug without being digested.

The single phase oily composition made of monoglyceride and oil or the composition of monoglyceride, oil and emulsifier have not been used as oral or buccal drug delivery systems in the past. Compositions including oils and water have been used for oral delivery. In these cases, the compositions form an L2 phase, in which small water droplets are formed inside the oil phase. The drugs are loaded inside and released from the water droplets. These compositions in L2 phase have many shortcomings

when compared to the compositions of the present invention. Once water is introduced into the system, the components can become destabilized due to oxidation and/or hydrolysis. Also the insoluble drugs can precipitate out with time. Also the administration dose would increase as the amount of 5 added water increases.

When the composition of the present invention includes an emulsifier, the emulsifier can help the composition to be dispersed inside the intestine into microparticles with the diameter of a few micrometers since the movement of intestine would help micronization process. Therefore, it is 10 possible to coat as wide area of the intestinal wall in case the composition includes an emulsifier.

The present inventors have proven that the composition containing monoglycerides and oils can solubilize the insoluble drugs, help preventing the precipitation of the drugs, and can be dispersed into microparticles in 15 water, can be adsorbed into the intestinal wall and therefore can increase oral bioavailability when orally consumed.

#### < Summary of the Invention >

The object of the present invention is to provide a composition for 20 solubilization of insoluble drugs and the preparation method thereof. Another object of the present invention is to provide a formulation by adding drugs in the above composition for solubilization of insoluble drugs to be used as drug delivery systems and the preparation method thereof.

**[DETAILED DESCRIPTION OF THE INVENTION]**

The present invention relates to a solubilizing composition of homogeneous oily mixture comprising monoglyceride and oil for 5 solubilization of insoluble drugs, and the preparation method thereof.

Also, the present invention relates to a novel formulation comprising the above solubilizing composition and pharmaceutical compounds, and the preparation method thereof.

Also the present invention relates to a solubilizing composition 10 including emulsifiers for solubilization of insoluble drugs comprising the above solubilizing composition and emulsifiers.

And the present invention also relates to another novel formulation comprising the above solubilizing composition including emulsifier and pharmaceutical compounds, and the preparation method thereof.

15 In what follows, the present invention will be described in detail.

The present invention relates to a mucoadhesive composition for solubilization of insoluble drugs.

Specifically, the above composition is composed of 4 ~ 90 % by weight of at least one selected from the monoglycerides and 0.01 ~ 90 % by 20 weight of at least one oil (with respect to the total weight of the composition).

The above composition can be prepared by mixing at least one monoglyceride and at least one oil at room or elevated temperatures.

The above monoglycerides are selected from a group consisting of one or more saturated or unsaturated monoglycerides having 10 ~ 22 carbon

atoms in the hydrocarbon chain. Monoglycerides is selected preferably from a group consisting of monoolein, monopalmitolein, monomyristolein, monelaidin and monoerucin and from a group consisting of the mixture of monoglycerides semi-synthesized from triglycerides of vegetable or animal oil, and more preferably monoolein.

The above oil is selected preferably from a group consisting of triglycerides, iodinated oil and vegetable or animal oil.

The above triglycerides are selected from a group consisting of one or more saturated or unsaturated triglycerides having 2 ~ 20 carbon atoms in the hydrocarbon chain. For instance, triacetin, tributyrin, tricaproin, tricaprylin, tricaprin or triolein can be used.

The above iodized oils include iodized poppy seed oil such as Lipiodol, Ethiodol and iodized soybean oil.

The above vegetable oils include soybean oil, cottonseed oil, olive oil, poppyseed oil, linseed oil and sesame oil.

The above animal oils include squalane and squalene.

Also the above composition can additionally include other additives up to 5 % by weight. For instance, the composition can further comprise alcohol, polyol or Cremophor to improve the solubility of the insoluble drugs, tocopherol or tocopherol acetate to prevent oxidation, and fatty acid, fatty acid ester or fatty acid alcohol to increase drug absorption.

The above solubilizing composition can be prepared by mixing 4 ~ 90 % by weight of at least one selected from the monoglycerides and 0.01 ~ 90 % by weight of at least one oil at temperatures lower than 50 °C to obtain

a homogeneous mixture. The monoglycerides and oils used in preparing the solubilizing composition are the same as described above.

The preparation method described above is only one of many possible methods, and other preparation method can also be used to obtain  
5 the above composition.

Also the present invention provides mucoadhesive composition including emulsifiers for solubilization of insoluble drugs

More particularly, the above composition is composed of 4~90 % by weight of at least one selected from the monoglycerides, 0.01~90 % by  
10 weight, of at least one oil and 0.01~90 % by weight of at least one emulsifier (with respect to the total weight of the composition).

The above composition can be prepared by adding at least one monoglyceride, at least one oil and at least one emulsifier at room or elevated temperatures.

15 The above monoglycerides are selected from a group consisting of one or more saturated or an unsaturated monoglycerides having 10 ~ 22 carbon atoms in the hydrocarbon chain. Monoglycerides is selected preferably from a group consisting of monoolein, monopalmitolein, monomyristolein, monelaidin and monoerucin and from a group consisting  
20 of the mixture of monoglycerides semi-synthesized from triglycerides of vegetable or animal oil, and more preferably monoolein.

The above oil is selected preferably from a group consisting of triglycerides, iodinated oil and vegetable or animal oil that can solubilize insoluble drugs.

The above triglycerides are selected from a group consisting of one or more saturated or unsaturated triglycerides having 2 ~ 20 carbon atoms in the hydrocarbon chain. For instance, triacetin, tributyrin, tricaproin, tricaprylin, tricaprin or triolein can be used.

5 The above iodized oils include iodized poppy seed oil such as Lipiodol, Ethiodol and iodized soybean oil.

The above vegetable oils include soybean oil, cottonseed oil, olive oil, poppyseed oil, linseed oil and sesame oil.

The above animal oils include squalane and squalene.

10 The emulsifier is selected from the group consisting of a phospholipid, a non-ionic surfactant, an anionic surfactant, a cationic surfactant, and a bile acid.

The phospholipid is selected from the group consisting of a phosphatidylcholine (PC) and its derivative, a phosphatidylethanolamine 15 (PE) and its derivative, a phosphatidylserine (PS) and its derivative or a polymeric lipid wherein a hydrophilic polymer is conjugated to the lipid headgroup.

The non-ionic surfactant is selected from the group consisting of a poloxamer (also known as Pluronic: polyoxyethylene-polyoxypropylene 20 copolymer), a sorbitan ester (Span), a polyoxyethylene sorbitan (Tween) or a polyoxyethylene ether (Brij).

The anionic surfactant is selected from the group consisting of a phosphatidylserine (PS) and its derivative, a phosphatidic acid (PA) and its derivative or sodium dodecyl sulfate (SDS).

The cationic surfactant is selected from the group consisting of 1,2-dioleyl-3-trimethylammonium propane (DOTAP), dimethyldioctadecylammonium bromide (DDAB), N-[1-(1,2-dioleyloxy)propyl]-N,N,N-trimethylammonium chloride (DOTMA),  
5 1,2-dioleyl-3-ethylphosphocholine (DOEPC) and 3β-[N-[(N',N'-dimethylamino)ethan]carbamoyl]cholesterol (DC-Chol).

The bile acid is selected from the group consisting of cholic acid, its salt and derivatives; deoxycholic acid, its salt and derivatives; chenocholic acid, its salt and derivatives; and lithocholic acid, its salt and derivatives.

10 Other additives can be added to the above composition including emulsifiers to be within 5% by weight. And the examples are fatty acids, fatty acid esters and fatty acid alcohols (with respect to the total weight of the composition). For instance, the composition can further comprise alcohol, polyol or Cremophor to improve the solubility of the insoluble drugs,  
15 tocopherol or tocopherol acetate to prevent oxidation, and fatty acid, fatty acid ester or fatty acid alcohol to increase drug absorption.

The above solubilizing composition including emulsifiers can be prepared by mixing 4 ~ 90 % by weight of at least one selected from the monoglycerides, 0.01 ~ 90 % by weight of at least one oil and 0.01 ~ 90 % by  
20 weight of at least one emulsifier at temperatures lower than 50 °C to obtain a homogeneous viscous mixture. The monoglycerides, oils and emulsifiers used in preparing the solubilizing composition are the same as described above.

25 The preparation method described above is only one of many possible methods, and other preparation method can also be used to obtain

the above composition including emulsifiers.

The compositions for solubilization of insoluble drugs with or without emulsifiers according to the present invention can be administered via various routes including oral administration, buccal administration, mucosal 5 administration, intranasal administration, intraperitoneal administration, subcutaneous injection, intramuscular injection, transdermal administration, intratumoral administration, and more preferably an oral administration.

The compositions for solubilization of insoluble drug of the present invention exist as gel or in semi-solid form depending on the composition at 10 room temperature. Also the compositions of the present invention are stable for a long period of time since the physical property of the composition does not change and the components do not degrade with time. Also the compositions for solubilization of insoluble drug of the present invention can be easily dispersed in water or in aqueous solutions to produce particles 15 bigger than 500 nm in diameter, and the absorbance of the dispersion at 400 nm is higher than 0.35 (preferably 1 ~ 4). Since the above dispersion of the compositions for solubilization of insoluble drug of the present invention does not form precipitation of the drug upon a long-time storage, the compositions of the present invention are efficient in solubilizing the insoluble drugs. 20 Since the compositions of the present invention are highly mucoadhesive in the intestine, they adhere onto absorptive cells in the intestine wherein the drug can be absorbed directly into the cells. The viscosity of the compositions is high enough (approximately 60 ~ 200 centipoises) to be adsorbed on a large area of the intestinal wall, thereby increasing the 25 amount of drug absorption per unit area. Another factor that helps increasing the bioavailability of drug in the composition for solubilization of

insoluble drug of the present invention is that it is composed of monoglycerides, which can be absorbed into the intestinal cells without being digested.

The present invention provides mucoadhesive formulations for  
5 solubilization of insoluble drugs that can be used as drug delivery systems.

Specifically, the above formulation is composed of 4 ~ 90 % by weight of at least one selected from the monoglycerides, 0.01 ~ 90 % by weight of at least one oil and 0.01 ~ 20 % by weight of insoluble drug (with respect to the total weight of the composition).

10 The above formulation can be prepared by mixing at least one monoglyceride, at least one oil and insoluble drug at room or elevated temperature.

15 The above monoglycerides are selected from a group consisting of one or more saturated or unsaturated monoglycerides having 10 ~ 22 carbon atoms in the hydrocarbon chain. Monoglyceride is selected preferably from a group consisting of monoolein, monopalmitolein, monomyristolein, monelaidin and monoerucin and from a group consisting of the mixture of monoglycerides semi-synthesized from triglycerides of vegetable or animal oil, and more preferably monoolein.

20 The above oil is selected preferably from a group consisting of triglycerides, iodinated oil and vegetable or animal oil.

25 The above triglycerides are selected from a group consisting of one or more saturated or unsaturated triglycerides having 2 ~ 20 carbon atoms in the hydrocarbon chain. For instance, triacetin, tributyrin, tricaproin, tricaprylin, tricaprin or triolein can be used.

The above iodized oils include iodized poppy seed oil such as Lipiodol, Ethiodol and iodized soybean oil.

The above vegetable oils include soybean oil, cottonseed oil, olive oil, poppyseed oil, linseed oil and sesame oil.

5       The above animal oils include squalane and squalene.

Examples of the insoluble drugs that can be used in the present invention are antivirals, steroid anti-inflammatory drugs (SAID), non-steroidal anti-inflammatory drugs (NSAID), antibiotics, antifungals, vitamins, hormones, retinoic acid, prostaglandins, prostacyclins, anticancer 10 drugs, antimetabolic drugs, miotics, cholinergics, adrenergic antagonists, anticonvulsants, antianxiety agents, major tranquilizers, antidepressants, anesthetics, analgesics, anabolic steroids, estrogens, progesterones, glycosaminoglycans, polynucleotides, immunosuppressants and immunostimulants.

15       Also the above formulation can additionally include other additives up to 5 % by weight. For instance, the composition can further comprise alcohol, polyol or Cremophor to improve the solubility of the insoluble drugs, tocopherol or tocopherol acetate to prevent oxidation, and fatty acid, fatty acid ester or fatty acid alcohol to increase drug absorption.

20       When applying these formulations in drug delivery system, it is preferred to use various administration routes including oral administration, buccal administration, mucosal administration, intranasal administration, intraperitoneal administration, subcutaneous injection, intramuscular injection, transdermal administration and intratumoral injection, and more 25 preferably an oral administration.

The preparation method of the above formulation for the solubilization of insoluble drugs comprises the steps of:

- 1) solubilizing 4 ~ 90 % by weight of at least one monoglyceride compounds in 0.01 ~ 90 % by weight of at least one oil at temperatures lower than 50 °C to obtain a homogeneous mixture (step 1); and
- 5  
2) solubilizing completely 0.01 ~ 20 % by weight of at least one insoluble drug in said mixture in step (1) (step 2).

The monoglycerides, oils and insoluble drugs used in preparing the 10 solubilizing formulation are the same as described above.

In step (2) of the above preparation method, the said mixture can be stirred or sonicated in a bath type sonicator to speed up the solubilization process.

Also the above formulation can be prepared by the following method 15 comprising the steps of:

- 1) mixing 4 ~ 90 % by weight of at least one monoglyceride compounds, 0.01 ~ 90 % by weight of at least one oil and 0.01 ~ 20 % of insoluble drug (step 1); and
- 20  
2) preparing a homogeneous liquid by solubilizing said mixture in step (1) completely (step 2).

The monoglycerides, oils and insoluble drugs used in preparing the solubilizing formulation are the same as above.

In step (2) of the above preparation method, the said mixture can be stirred or sonicated in a bath type sonicator at temperatures lower than 50 °C

to speed up the solubilization process.

The preparation methods described above are only examples of many possible methods, and other preparation method can also be used to obtain the above formulation for the solubilization of insoluble drug.

5       Also, the present invention provides the formulation for the solubilization of insoluble drug that uses the mucoadhesive composition including emulsifiers for solubilization of insoluble drugs as a drug delivery system.

More particularly, the above formulation is composed of 4 ~ 90% by  
10 weight of at least one selected from the monoglycerides, 0.01 ~ 90 % by weight of at least one oil, 0.01 ~ 90 % by weight of at least one emulsifier and 0.01 ~ 20 % by weight of insoluble drug (with respect to the total weight of the composition).

The above formulation can be prepared by adding at least one  
15 monoglyceride, at least one oil, at least one emulsifier and insoluble drug at room or elevated temperatures.

The above monoglycerides are selected from a group consisting of one or more saturated or unsaturated monoglycerides having 10 ~ 22 carbon atoms in the hydrocarbon chain. Monoglyceride is selected preferably from  
20 a group of consisting of monoolein, monopalmitolein, monomyristolein, monoelaidin and monoerucin, and semi-synthesized monoglycerides and their mixtures from triglycerides extracted from vegetable or animal oils, and more preferably monoolein.

The above oil solubilizing insoluble drugs is selected preferably from  
25 a group consisting of triglycerides, iodinated oil, vegetable oil or animal oil.

The above triglycerides are selected from a group consisting of one or more saturated or unsaturated triglycerides having 2 ~ 20 carbon atoms in the hydrocarbon chain. For instance, triacetin, tributyrin, tricaproin, tricaprylin, tricaprin or triolein can be used.

5 The above iodized oils include iodized poppy seed oil such as Lipiodol, Ethiodol and iodized soybean oil.

The above vegetable oils include soybean oil, cottonseed oil, olive oil, poppyseed oil, linseed oil and sesame oil.

The above animal oils include squalane and squalene.

10 The above emulsifier is selected from the group consisting of phospholipid, a non-ionic surfactant, an anionic surfactant, a cationic surfactant, and bile acid.

The phospholipid is selected from the group consisting of a phosphatidylcholine (PC) and its derivative, a phosphatidylethanolamine 15 (PE) and its derivative, a phosphatidylserine (PS) and its derivative and a polymeric lipid wherein a hydrophilic polymer is conjugated to the lipid headgroup.

20 The non-ionic surfactant is selected from the group consisting of a poloxamer (also known as Pluronic: polyoxyethylene-polyoxypropylene copolymer), a sorbitan ester (Span), a polyoxyethylene sorbitan (Tween) and a polyoxyethylene ether (Brij).

The anionic surfactant is selected from the group consisting of a phosphatidylserine (PS) and its derivative, a phosphatidic acid (PA) and its derivative and sodium dodecyl sulfate (SDS).

The cationic surfactant is selected from the group consisting of 1,2-dioleyl-3-trimethylammonium propane (DOTAP), dimethyldioctadecylammonium bromide (DDAB), N-[1-(1,2-dioleyloxy)propyl]-N,N,N-trimethylammonium chloride (DOTMA),  
5 1,2-dioleyl-3-ethylphosphocholine (DOEPC) or  
3 $\beta$ -[N-[(N',N'-dimethylamino)ethan]carbamoyl]cholesterol (DC-Chol).

The bile acid is selected from the group consisting of cholic acid, its salt and derivatives; deoxycholic acid, its salt and derivatives; chenocholic acid, its salt and derivatives; and lithocholic acid, its salt and derivatives.

10 The above insoluble drugs that can be used in the present invention are antivirals, steroid anti-inflammatory drugs (SAID), non-steroidal anti-inflammatory drugs (NSAID), antibiotics, antifungals, vitamins, hormones, retinoic acid, prostaglandins, prostacyclins, anticancer drugs, antimetabolitic drugs, miotics, cholinergics, adrenergic antagonists,  
15 anticonvulsants, antianxiety agents, major tranquilizers, antidepressants, anesthetics, analgesics, anabolic steroids, estrogens, progesterones, glycosaminoglycans, polynucleotides, immunosuppressants and immunostimulants

Other additives can be added to the above formulation including  
20 emulsifiers to be within 5% by weight. For instance, the composition can further comprise alcohol, polyol or Cremophor to improve the solubility of the insoluble drugs, tocopherol or tocopherol acetate to prevent oxidation, and fatty acid, fatty acid ester or fatty acid alcohol to increase drug absorption.

The formulations for solubilization of insoluble drugs with emulsifiers  
25 according to the present invention can be administered via various routes

including oral administration, buccal administration, mucosal administration, nasal administration, intraperitoneal administration, subcutaneous injection, intramuscular injection, transdermal administration, intratumoral administration, and more preferably an oral administration.

5       The preparation method of the above formulation for the solubilization of insoluble drugs that uses the mucoadhesive composition including emulsifiers for solubilization of insoluble drugs as a drug delivery system comprises the steps of:

- 1) preparing a viscous liquid by solubilizing completely 4 ~ 90 % by weight of at least one monoglyceride compound, 0.01 ~ 90 % by weight of at least one oil and 0.01 ~ 90 % by weight of at least one emulsifier at temperatures lower than 50 °C to obtain a homogeneous mixture (step 1); and
- 10      2) preparing a homogeneous liquid formulation by mixing 0.01 ~ 20 % by weight of at least one insoluble drug in said liquid in step 15 (1) (step 2).

The monoglycerides, oils, emulsifiers and insoluble drugs used in preparing the solubilizing formulation are the same as described above.

For example, after adding insoluble drug in a viscous liquid obtained 20 by mixing completely monoglyceride compound, oil and emulsifier, the mixture can be stirred or sonicated for 3 ~ 5 minutes at room temperature or temperatures lower than 50 °C to speed up the solubilization process.

Another preparation method of the above formulation for the solubilization of insoluble drugs that uses the mucoadhesive composition 25 including emulsifiers for solubilization of insoluble drugs as a drug delivery

system comprises the steps of:

- 1) preparing viscous liquid containing insoluble drug by mixing 0.01 ~ 90 % by weight of at least one oil and 0.01 ~ 20 % by weight of insoluble drug and sonicating in a bath type sonicator (step 1);  
5 and
- 2) preparing a homogeneous liquid formulation by mixing 0.01 ~ 90 % by weight of at least one emulsifier and 4 ~ 90 % by weight of at least one monoglyceride in said liquid in step (1) (step 2).

The monoglycerides, oils, emulsifiers and insoluble drugs used in  
10 preparing the solubilizing formulation are the same as above.

The preparation methods described above is only two of many possible methods, and other preparation methods can also be used to obtain the above formulation that uses the mucoadhesive composition including emulsifiers for solubilization of insoluble drugs as a drug delivery system.

15 The compositions for solubilization of insoluble drug of the present invention exist as liquid or in semi-solid state depending on the temperature at which they exist. The physical state of the formulation depends on the melting point. In general, the formulation exists as semi-solid at room temperature (ca. 25 °C), and as liquid at temperatures above room  
20 temperature. Also the melting point of the above formulation depends on the kinds and the amount of the additives. One of the general characteristics of the formulations is that they exist as a viscous liquid at body temperature and can be adsorbed on a wide area of the intestine.

The formulations of the present invention in viscous liquid, gel or

semi-solid form are stable for a long period of time since the physical property of the composition does not change and the components including the insoluble drug do not degrade with time. Also the formulations for solubilization of insoluble drug of the present invention is an efficient 5 solubilization system of the insoluble drugs since they can be easily dispersed in water or in aqueous solutions to produce particles bigger than 300 nm in diameter, and the dispersion does not form aggregates with time.

#### BRIEF DESCRIPTION OF THE DRAWINGS

10

Figure 1 is a graph showing the concentration of pyrene in blood and in different organs after oral administration of the liquid formulation solubilizing insoluble drug in Example 20 of the present invention. The quantitative analysis of pyrene was performed by HPLC. Tricaprylin 15 emulsion including pyrene was orally administered as a control group.

- ■ - ; a group orally administered with liquid formulation for solubilization of insoluble drugs of the present invention (2 mg pyrene, weight ratio of the liquid formulation is monoolein:tricaprylin:pyrene = 64.3:32.2:3.5, the liquid formulation in Example 20),
- 20 - □ - ; a group orally administered with tricaprylin emulsion (2 mg pyrene, weight ratio of the emulsion is tricaprylin:Tween 80:pyrene:water = 8.65:0.95:0.4:90, the control group in Example 25).

Figure 2 is a graph showing the concentration of pyrene in the intestine 1 or 2 hours after oral administration of the liquid formulation

solubilizing insoluble drug in Example 20 of the present invention. The quantitative analysis of pyrene was performed by HPLC. Tricaprylin emulsion including pyrene was orally administered as a control group.

Test group: a group orally administered with liquid formulation for  
5 solubilization of insoluble drugs of the present invention (2 mg pyrene,  
weight ratio of the liquid formulation is monoolein:tricaprylin:pyrene =  
64.3:32.2:3.5, the liquid formulation in Example 20),

Control group: a group orally administered with tricaprylin emulsion  
(2 mg pyrene, weight ratio of the emulsion is tricaprylin:Tween  
10 10:80:pyrene:water = 8.65:0.95:0.4:90, the control group in Example 25).

- ■ - ; pyrene concentration in the intestine 1 hour after oral  
administration,

- □ - ; pyrene concentration in the intestine 2 hours after oral  
administration.

15 Figure 3 is a graph showing the concentration of pyrene in blood and  
in different organs after oral administration of the liquid formulation including  
emulsifier solubilizing insoluble drug in Example 24 of the present invention.  
The quantitative analysis of pyrene was performed by HPLC. Tricaprylin  
emulsion including pyrene was orally administered as a control group.

20 - ■ - ; a group orally administered with liquid formulation for  
solubilization of insoluble drugs of the present invention (2 mg pyrene,  
weight ratio of the liquid formulation is monoolein:tricaprylin:Tween 80:pyrene  
= 53.6:26.8:16.1:3.5, the liquid formulation in Example 24),

- □ - ; a group orally administered with tricaprylin emulsion (2 mg

pyrene, weight ratio of the emulsion is tricaprylin:Tween 80:pyrene:water = 8.65:0.95:0.4:90, the control group in Example 25).

Figure 4 is a graph showing the concentration of pyrene in the intestine 1 or 2 hours after oral administration of the liquid formulation including emulsifier solubilizing insoluble drug in Example 24 of the present invention. The quantitative analysis of pyrene was performed by HPLC. Tricaprylin emulsion including pyrene was orally administered as a control group.

Test group: a group orally administered with liquid formulation for solubilization of insoluble drugs of the present invention (2 mg pyrene, weight ratio of the liquid formulation is monoolein:tricaprylin:Tween 80:pyrene = 53.6:26.8:16.1:3.5, the liquid formulation in Example 24),

Control group: a group orally administered with tricaprylin emulsion (2 mg pyrene, weight ratio of the emulsion is tricaprylin:Tween 80:pyrene:water = 8.65:0.95:0.4:90, the control group in Example 25).

- ■ - ; pyrene concentration in the intestine 1 hour after oral administration,

- □ - ; pyrene concentration in the intestine 2 hours after oral administration.

### [Best Mode for Carrying Out the Invention]

This invention is explained in more detail based on the following Examples but they should not be construed as limiting the scope of this

invention.

**Example 1. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the 5 composition (1)**

① Manufacturing mucoadhesive composition for solubilization of insoluble drugs

A mucoadhesive composition for solubilization of insoluble drugs, which is a viscous oily solution, was prepared by mixing 1 g monoolein and 10 0.5 g tricaprylin and warmed at 40 °C. Monoolein used in Examples 1 and below was Myverol 18-99 K from Danisco A/S (Copenhagen, Denmark) with the monoolein content of 86.6 weight %.

② Property Analysis of thus prepared solubilizing composition

The size of the emulsion particles were measured by using Malvern 15 Zetasizer (Malvern Instruments Limited, England) after preparing the emulsion by adding 3 mL of distilled water to 2 µL of thus obtained liquid formulation. An average particle size and polydispersity were obtained by measuring values for a given formulation three times (Orr, *Encyclopedia of 20 emulsion technology*, 1, 369-404, 1985). The polydispersity was obtained as the variance indicated by the logarithmic scale in the logarithmic normal distribution function. The above method in measuring the particle size and the polydispersity was used throughout the following examples.

The above composition exists as semi-solid or solid at room temperature and in a refrigerator, respectively, but as liquid at or above 40

°C. Dispersion with the average particle size of 530 nm was obtained when the above composition was vortexed for 10 s in water. The absorbance at 400 nm was 2.36.

5       **Example 2. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the composition (2)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein and 1 g tricaprylin were used, and their 10 particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 730 nm was obtained. The absorbance at 400 nm was 2.23.

15       **Example 3. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the composition (3)**

The composition was prepared by the same methods in Example 1 with the exception that 0.5 g monoolein and 1 g tricaprylin were used, and their particle size and polydispersity were measured by the same methods in 20 Example 1. Dispersion with the average particle size of 554 nm was obtained. The absorbance at 400 nm was 2.54.

The results of the Examples 1-3 are summarized in the following Table 1.

**Table 1**

Content (weight %)		Particle size (nm) (polydispersity)	Absorbance (400 nm)	Example
Monoolein	Tricaprylin			
66.7	33.3	530 (1)	2.36	1
50	50	730 (1)	2.23	2
33.3	66.7	554 (0.683)	2.54	3

### **Comparative Example 1**

#### **Dispersion of monoolein**

Monoolein (99.5 % purity) from Nu-Chek Prep (Elysian, MN, USA) or 5 Myverol 18-99 K (monoolein content 86.6 weight %) from Danisco A/S (Copenhagen, Denmark) were mixed with water. Cubic phases were formed instead of dispersion. Since the cubic phase had very high viscosity and floated in water, the particle size or absorbance could not be determined.

10           **Example 4. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the oil and the composition (1)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein and 0.5 g tributyrin were used. The 15 particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 303 nm was obtained. The absorbance at 400 nm was 0.78.

20           **Example 5. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the oil and the composition (2)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein and 1 g tributyrin were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 319 nm was obtained.

5 The absorbance at 400 nm was 0.37.

**Example 6. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the oil and the composition (3)**

10 The composition was prepared by the same methods in Example 1 with the exception that 0.5 g monoolein and 1 g tributyrin were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 916 nm was obtained. The absorbance at 400 nm was 2.19.

15 The results of the Examples 4-6 are summarized in the following Table 2.

**Table 2**

Content (weight %)		Particle size (nm) (polydispersity)	Absorbance (400 nm)	Example
Monoolein	Tributyrin			
66.7	33.3	303 (0.246)	0.78	4
50	50	319 (0.255)	0.37	5
33.3	66.7	916 (1)	2.19	6

20 **Example 7. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the oil (1)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein and 0.5 g squalane were used. The particle size and polydispersity were measured by the same methods in Example 1. An unstable dispersion with the average particle size of 1570 nm 5 was obtained. The absorbance at 400 nm was 2.48.

**Example 8. Mucoadhesive composition for solubilization of insoluble drugs manufactured according to the change in the oil (2)**

The composition was prepared by the same methods in Example 1 10 with the exception that 1 g monoolein and 0.5 g lipiodol (Lipiodol Ultra-fluid, Laboratoire Guerbet, France, Iodine content: 38 % by weight) were used. The particle size and polydispersity were measured by the same methods in Example 1. An unstable dispersion with the average particle size of 245 nm 15 was obtained. The absorbance at 400 nm was 0.57.

15 The results of the Examples 7 and 8 are summarized in the following Table 3.

**Table 3**

Oil *	Particle size (nm) (polydispersity)	Absorbance (400 nm)	Example
Squalane	1570 (1.000)	2.48	7
Lipiodol	245 (0.158)	0.57	8
*Monoolein:Oil = 66.7: 33.3 (Weight ratio)			

20 **Example 9. Mucoadhesive composition including emulsifiers for solubilization of insoluble drugs manufactured according to the change in the composition (1)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein, 0.5 g tricaprylin and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 583 nm 5 was obtained. The absorbance at 400 nm was 2.68.

**Example 10. Mucoadhesive composition including emulsifiers for solubilization of insoluble drugs manufactured according to the change in the composition (2)**

10 The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein, 1 g tricaprylin and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 397 nm was obtained. The absorbance at 400 nm was 0.94.

15

**Example 11. Mucoadhesive composition including emulsifiers for solubilization of insoluble drugs manufactured according to the change in the composition (3)**

The composition was prepared by the same methods in Example 1  
20 with the exception that 0.5 g monoolein, 1 g tricaprylin and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 587 nm was obtained. The absorbance at 400 nm was 1.32.

The results of the Examples 9-11 are summarized in the following

Table 4.

**Table 4**

Content (weight %)			Particle size (nm) (polydispersity)	Absorbance (400 nm)	Example
Monoolein	Tricaprylin	Tween 80			
53.3	26.7	20	583 (1)	2.68	9
40	40	20	397 (0.605)	0.94	10
26.7	53.3	20	587 (0.211)	1.32	11

**Example 12. Mucoadhesive composition including emulsifiers  
5 for solubilization of insoluble drugs manufactured according to the  
change in the oil and the composition (1)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein, 0.5 g tributyrin and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same 10 methods in Example 1. Dispersion with the average particle size of 1168 nm was obtained. The absorbance at 400 nm was 2.35.

**Example 13. Mucoadhesive composition including emulsifiers  
15 for solubilization of insoluble drugs manufactured according to the  
change in the oil and the composition (2)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein, 1 g tributyrin and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same 20 methods in Example 1. Dispersion with the average particle size of 170 nm was obtained. The absorbance at 400 nm was 0.41.

**Example 14. Mucoadhesive composition including emulsifiers for solubilization of insoluble drugs manufactured according to the change in the oil and the composition (3)**

5       The composition was prepared by the same methods in Example 1 with the exception that 0.5 g monoolein, 1 g tributyrin and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 650 nm was obtained. The absorbance at 400 nm was 2.56.

10      The results of the Examples 12-14 are summarized in the following Table 5.

**Table 5**

Content (weight %)			Particle size (nm) (polydispersity)	Absorbance (400 nm) Monoolein	Example
Monoolein	Tributyrin	Tween 80			
53.3	26.7	20	1168 (1)	2.35	12
40	40	20	170 (0.946)	0.41	13
26.7	53.3	20	650 (0.863)	2.56	14

15      **Example 15. Mucoadhesive composition including emulsifiers for solubilization of insoluble drugs manufactured according to the change in the oil (1)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein, 0.5 g squalane and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 506 nm

was obtained. The absorbance at 400 nm was 1.75.

**Example 16. Mucoadhesive composition including emulsifiers  
for solubilization of insoluble drugs manufactured according to the  
5 change in the oil (2)**

The composition was prepared by the same methods in Example 1 with the exception that 1 g monoolein, 0.5 g lipiodol and 0.3 g Tween 80 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 913 nm  
10 was obtained. The absorbance at 400 nm was 3.10.

The results of the Examples 15-16 are summarized in the following  
Table 6.

**Table 6**

Oil *	Particle size (nm) (polydispersity)	Absorbance (400 nm) Monoolein	Example
Squalane	506 (0.407)	1.75	15
Lipiodol	913 (0.472)	3.10	16

\*Monoolein:Oil:Tween 80 = 53.3: 26.7:20 (Weight ratio)

**15 Example 17. Preparation of mucoadhesive formulation for  
solubilization of insoluble drugs (1)**

A mucoadhesive formulation for solubilization of insoluble drugs, which is a viscous oily solution, was prepared by mixing 1 g monoolein, 0.5 g tricaprylin and 15 mg cyclosporine A, an insoluble drug and warmed at 40 °C,  
20 and their particle size and polydispersity were measured by the same

methods in Example 1. Dispersion with the average particle size of 1525 nm was obtained. The absorbance at 400 nm was 1.39.

**Example 18. Preparation of mucoadhesive formulation for  
5 solubilization of insoluble drugs (2)**

A mucoadhesive formulation for solubilization of insoluble drugs was prepared by the same methods in Example 17 with the exception that 1 g monoolein, 0.5 g tricaprylin and 15 mg felodipin, an insoluble drug were used. The particle size and polydispersity were measured by the same methods in  
10 Example 1. Dispersion with the average particle size of 953 nm was obtained. The absorbance at 400 nm was 1.85.

The results of the Examples 17 and 18 are summarized in the following Table 7.

**Table 7**

Drug*	Particle size (nm) (polydispersity)	Absorbance (400 nm)	Example
Cyclosporin A	1525 (1)	1.39	17
Felodipin	953 (1)	1.85	18

\*Monoolein:Tricaprylin:Drug = 66:33:1 (Weight ratio)

15

**Example 19. Preparation of mucoadhesive formulation for  
solubilization of insoluble drugs (3)**

Mucoadhesive formulations for solubilization of insoluble drugs were prepared by mixing 1 g of the composition prepared in Examples 1 through 8  
20 and 0.4 mg pyrene and warmed at 40 °C, and their particle size and polydispersity were measured by the same methods in Example 1.

The results of the Example 19 are summarized in the following Table 8.

Table 8

Content (weight %)			Content (weight %)	Absorbance (400 nm)	Example
Monoolein	Tricaprylin	Pyrene			
66.64	33.32	0.04	628 (1)	2.19	1
49.98	49.98	0.04	729 (1)	2.05	2
33.32	66.64	0.04	533 (1)	2.48	3
Monoolein	Tributyrin	Pyrene			
66.64	33.32	0.04	503 (1)	2.42	4
49.98	49.98	0.04	555 (1)	2.47	5
33.32	66.64	0.04	698 (1)	2.46	6
Monoolein	Squalene	Pyrene			
66.64	33.32	0.04	963 (1)	2.70	7
Monoolein	Lipiodol	Pyrene			
66.64	33.32	0.04	246 (0.137)	0.63	8

5       **Example 20. Preparation of mucoadhesive formulation for solubilization of insoluble drugs (4)**

A mucoadhesive formulation for solubilization of insoluble drugs was prepared by the same methods in Example 17 with the exception that 1 g monoolein, 0.5 g tricaprylin and 55 mg pyrene, an insoluble model drug, 10 were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 738 nm was obtained. The absorbance at 400 nm was 2.35.

**Example 21. Preparation of mucoadhesive formulation including**

### **emulsifiers for solubilization of insoluble drugs (1)**

A mucoadhesive formulation including emulsifiers for solubilization of insoluble drugs, which is a viscous oily solution, was prepared by mixing 1 g monoolein, 0.5 g tricaprylin, 0.3 mg Tween 80 and 18 mg cyclosporine A, an 5 insoluble drug, were used. The particle size and polydispersity were measured by the same methods in Example 1. Dispersion with the average particle size of 1940 nm was obtained. The absorbance at 400 nm was 2.13.

### **Example 22. Preparation of mucoadhesive formulation including 10 emulsifiers for solubilization of insoluble drugs (2)**

A mucoadhesive formulation for solubilization of insoluble drugs was prepared by the same methods in Example 20 with the exception that 1 g monoolein, 0.5 g tricaprylin, 0.3 g Tween 80 and 18 mg felodipin, an insoluble drug, were used. The particle size and polydispersity were 15 measured by the same methods in Example 1. Dispersion with the average particle size of 838 nm was obtained. The absorbance at 400 nm was 2.63.

The results of the Examples 21 and 22 are summarized in the following Table 9.

**Table 9**

20

Drug*	Particle size (nm) (polydispersity)	Absorbance (400 nm) Monoolein	Example
Cyclosporin A	1940 (1)	2.13	21
Felodipin	838 (1)	2.63	22

\*Monoolein:Tricaprylin:Tween 80:Drug = 55:28:16:1 (Weight ratio)

**Example 23. Preparation of mucoadhesive formulation including emulsifiers for solubilization of insoluble drugs (3)**

Mucoadhesive formulations for solubilization of insoluble drugs was prepared by mixing 1 g of the composition prepared in Examples 9 through 16 and 0.4 mg pyrene and warmed at 40 °C, and their particle size and polydispersity were measured by the same methods in Example 1.

The results of the Example 19 are summarized in the following Table 10.

10

**Table 10**

Content (weight %)				Particle size (nm) (polydispersity)	Absorbance (400 nm)	Example
Monoolein	Tricaprylin	Tween 80	Pyrene			
53.31	26.66	19.99	0.04	668 (1)	2.85	9
39.985	39.985	19.99	0.04	517 (1)	2.74	10
26.66	53.31	19.99	0.04	764 (0.477)	2.92	11
Monoolein	Tributyrin	Tween 80	Pyrene			
53.31	26.66	19.99	0.04	721 (1)	2.61	12
39.985	39.985	19.99	0.04	526 (1)	2.89	13
26.66	53.31	19.99	0.04	588 (1)	2.82	14
Monoolein	Squalene	Tween 80	Pyrene			
53.31	26.66	19.99	0.04	400 (0.254)	1.35	15
Monoolein	Lipiodol	Tween 80	Pyrene			
53.31	26.66	19.99	0.04	643 (0.739)	3.28	16

**Example 24. Preparation of mucoadhesive formulation including emulsifiers for solubilization of insoluble drugs (4)**

A mucoadhesive formulation for solubilization of insoluble drugs was prepared by the same methods in Example 20 with the exception that 1 g monoolein, 0.5 g tricaprylin, 0.3 g Tween 80 and 65.3 mg pyrene, an insoluble model drug, were used. The particle size and polydispersity were 5 measured by the same methods in Example 1. Dispersion with the average particle size of 698 nm was obtained. The absorbance at 400 nm was 2.93.

**Example 25. *In vivo* oral administration of mucoadhesive formulation for solubilization of insoluble drugs**

10 Animal experiments were performed by using the mucoadhesive formulations for the solubilization of insoluble drugs prepared in the above Example 20.

① Oral Administration of mucoadhesive formulations for the solubilization of insoluble drugs

15 Fifty-six microliters of the mucoadhesive formulation containing 2 mg pyrene was administered into Balb/C mouse (6 ~ 7 weeks old, female) fasted for 4 hours previously by using a gastric sonde. Tricaprylin emulsion containing pyrene was prepared as a control group. Tricaprylin emulsion was prepared by mixing tricaprylin, tween 80 and pyrene at a 20 weight ratio of 86.5: 9.5: 4 and solubilized completely by heating the mixture to 50 °C. One milliliter of the mixture was mixed with 9 ml water and sonicated for 2 min by using a probe type sonicator (High intensity ultrasonic processor, microprocessor control, 600-Watt model). The particle size and the polydispersity of the prepared emulsion were 103 nm 25 and 0.2, respectively, and the absorbance at 400 nm was 0.3. Tricaprylin

emulsion (500 µl) containing 2 mg pyrene was administered orally for comparison. One, 2, 3, 4 and 6 h after the oral administration of the formulations, the concentrations of pyrene in the blood and in various organs were determined.

5

② Determination of pyrene concentration in blood and in various organs

Blood withdrawn and organs taken out from the animal were mixed with methanol (8 folds in weight) and centrifuged at 14000 rpm at 4 °C for 15 min. After the mixture was centrifuged, the supernatant was taken to determine the concentration of pyrene by Fluorimetry ( $\lambda_{\text{ex}} = 336 \text{ nm}$ ,  $\lambda_{\text{em}} = 389 \text{ nm}$ ). The concentrations of pyrene one hour after oral administration in each organ and in blood are shown in Figure 1. It is well known that insoluble chemicals like pyrene can be absorbed into the intestinal cells when solubilized in hydrophobic particles, such as tricaprylin emulsion. It is notable that pyrene solubilized in the viscous liquid formulation of the present invention can be also absorbed into the body. Also the concentration of pyrene in the intestine increases with time similar to the case of tricaprylin emulsion control group as shown in Figure 2.

20

**Example 26. *In vivo* oral administration of mucoadhesive formulation including emulsifiers for solubilization of insoluble drugs 1**

Animal experiments were performed by using the mucoadhesive formulations for the solubilization of insoluble drugs prepared in the above

Example 24.

① Oral Administration of mucoadhesive formulations for the solubilization of insoluble drugs

Fifty-six microliters of the mucoadhesive formulation containing 2 mg pyrene was administered into Balb/C mouse (6 ~ 7 weeks old, female) fasted for 4 hours previously by using a gastric sonde. Tricaprylin emulsion containing pyrene was prepared and orally administered as a control group as in Example 25. One and two hours after the oral administration of the formulations, the concentrations of pyrene in the blood and in various organs were determined.

② Determination of pyrene concentration in blood and in various organs

The concentrations of pyrene one hour after oral administration in each organ and blood were quantified as in Example 25 and the result is shown in Figure 3. It is well known that insoluble chemicals like pyrene can be absorbed into the intestinal cells when solubilized in hydrophobic particles, such as tricaprylin emulsion. It is notable that pyrene solubilized in a viscous liquid formulation of the present formulation can be absorbed into the body. Also the concentration of pyrene in the intestine increases with time higher than that in the case of tricaprylin emulsion control group as shown in Figure 4.

**[Industrial Applicability]**

As described above, the mucoadhesive composition for solubilization of insoluble drugs according to the present invention can solubilize insoluble drug stably and also does not form precipitates of insoluble drug when dispersed in water. Since the mucoadhesive composition for solubilization of insoluble drugs according to the present invention can encapsulate and increase the absorption of insoluble drugs efficiently, it is suitable for oral and intraperitoneal delivery, and can be efficiently perish tumor cells.